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Education

Supplementary Material

PDF version of File 3. MATHCAD simulation protocol of EC' mechanism

FROM THEORY TO SIMULATION: OPEN INTERACTIVE MATHCAD SIMULATION PROTOCOLS FOR EXPLORING COMMON ELECTRODE MECHANISMS IN CYCLIC VOLTAMMETRY

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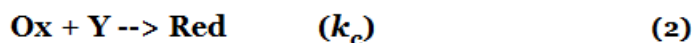
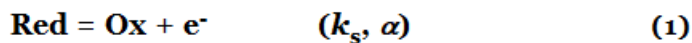
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EC' electrode mechanism at a planar electrode of a dissolved redox couple in Cyclic Staircase Voltammetry



$E_s := -0.5$ starting potential (in V vs. the formal potential)

$E_f := 0.5$ switching potential (in V vs. the formal potential)

$dE := 0.005$ potential step increment (in V)

$\Delta E := E_f - E_s$ potential window

$v := 0.1$ potential scan rate in V/s

$\tau := \frac{dE}{v}$ duration of a single step (in s)

$\tau = 0.05$

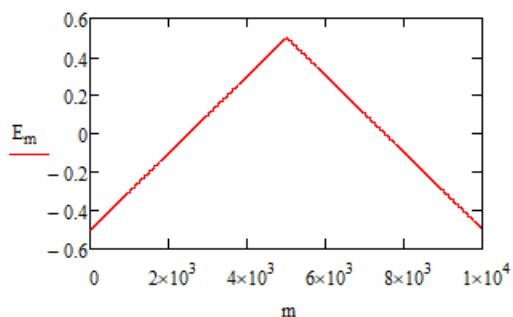
$M := 25$ number of time increments in a single potential step

$d := \frac{\tau}{M}$ time increment (in s)

$2 \cdot \frac{\Delta E}{dE} = 400$ total number of potential steps

$m_{\text{max}} := 1..2 \cdot \frac{\Delta E}{dE} \cdot 25$ serial number of time increments

$$E_m := \text{if} \left[m \leq \frac{\Delta E}{dE} \cdot 25, E_s + \left(\text{ceil} \left(\frac{m}{25} \right) \cdot dE - dE \right), E_f - \left[\text{ceil} \left[\frac{m - \left(\frac{\Delta E}{dE} \cdot 25 \right)}{25} \right] \cdot dE - dE \right] \right] \quad \text{potential ramp} \quad (3)$$



$F := 96485$ Faraday constant in C/mol
 $T := 298.15$ thermodynamic temperature in K

$R := 8.314$ Gass constant in J/(mol K)

$n := 1$ stoichiometric number of electrons

$\Phi_m := n \cdot \frac{F}{R \cdot T} \cdot E_m$ dimensionless potential (4)

$D := 5 \cdot 10^{-6}$ common diffusion coefficient in cm²/s

$k_s := 0.005$ electrochemical standard rate constant in cm/s

$\alpha := 0.5$ electron transfer coefficient

$k_c := 0.6$ rate constant of the chemical regenerative recation in s⁻¹

$K := \frac{k_s \cdot \sqrt{\tau}}{\sqrt{D}}$ dimensionless electrode kinetic parameter

$K_{chem} := k_c \cdot \tau$ dimensionless chemical kinetic parameter

$$s_{\infty m} := \sqrt{m} - \sqrt{m-1} \quad \text{numerical integration parameter} \quad (5)$$

$$M_{\infty m} := \operatorname{erf}\left(\sqrt{K_{\text{chem}} \cdot \frac{m}{25}}\right) - \operatorname{erf}\left[\sqrt{K_{\text{chem}} \cdot \frac{(m-1)}{25}}\right] \quad \text{numerical integration parameter} \quad (6)$$

$$\Psi_1 := \frac{K \cdot e^{-\alpha \cdot \Phi_1}}{1 + \frac{K \cdot e^{-\alpha \cdot \Phi_1} \cdot M_1}{\sqrt{K_{\text{chem}}}} (1 + e^{-\Phi_1})} \quad (7)$$

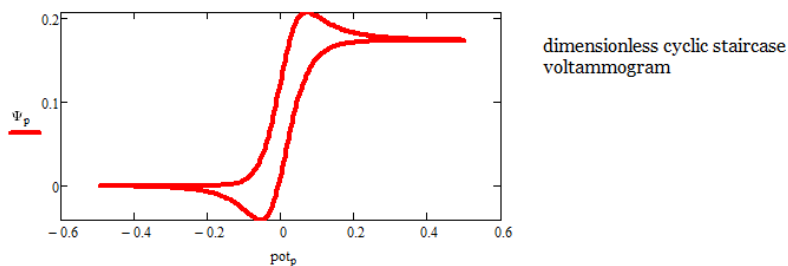
Recurrent formulas for calculating the dimensionless current

$$\Psi_m := \frac{K \cdot e^{-\alpha \cdot \Phi_m} \left[1 - \frac{1 + e^{-\Phi_m}}{\sqrt{K_{\text{chem}}}} \sum_{j=1}^{m-1} (\Psi_j \cdot M_{m-j+1}) \right]}{1 + \frac{K \cdot e^{-\alpha \cdot \Phi_m} \cdot M_1}{\sqrt{K_{\text{chem}}}} (1 + e^{-\Phi_m})} \quad (8)$$

$$p := 1.2 \cdot \frac{\Delta E}{dE} - 1 \quad \text{serial number of potential steps} \quad (9)$$

$$\Psi_p := \Psi \left(\frac{\tau}{d \cdot 25} + p \right) \cdot 25 \quad \text{dimensionless current at the end of each potential step} \quad (10)$$

$$\text{pot}_p := \text{if} \left[p \leq \frac{\Delta E}{dE} \cdot E_s + p \cdot dE, E_f - \left(p - \frac{\Delta E}{dE} \right) \cdot dE \right] \quad \text{potential value of each potential step in V} \quad (11)$$



$s = 0.05$ electrode surface area in cm^2

$c_{\infty} = 1 \cdot 10^{-6}$ bulk concentration of the electroactive reactant in mol/cm^3

$A_{\infty} = n \cdot F \cdot S \cdot c_{\infty} \cdot \left(\frac{D}{\tau} \right)$ amperometric constant

$I_p = 10^6 \cdot \Psi_p \cdot A_{\infty}$ real current in μA

